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that can exist either alone or in molecules containing the

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CHAPTER 5 REVIEW The Periodic Law SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In the modern periodic table, elements are ordered (a) according to decreasing atomic mass. (b) according to Mendeleev's original design. (c) according to increasing atomic number. (d) based on when they were discovered. 2. d Mendeleev noticed that certain similarities in the ...

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Modern Chemistry 47 Chemical Bonding CHAPTER 6 REVIEW Chemical Bonding SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. \_\_\_\_\_ In metals, the valence electrons are considered to be (a) attached to particular positive ions.

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18. Dalton's atomic theory acyreed with the modern atomic theory EXCEPT for the statement that all atoms of the same element have the same Complete the following table to compare the types of subatomic particles. 19. 20. 21. 10 Particle Proton Neutron Electron CHAPTER 3 TEST Mass n mber Relative charge Locatio MODERN CHEMISTRY

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